

2020 NSF CYBERINFRASTRUCTURE FOR SUSTAINED SCIENTIFIC INNOVATION (CSSI) PRINCIPAL INVESTIGATOR MEETING

ELEMENTS: COMMUNITY PORTAL FOR HIGH-PRECISION ATOMIC PHYSICS DATA AND COMPUTATION

PI: Marianna Safronova, Co-PI: Rudolf Eigenmann, University of Delaware



University of Delaware project team and collaborators

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Award #1931339

Extraordinary progress in the control of atoms and ions

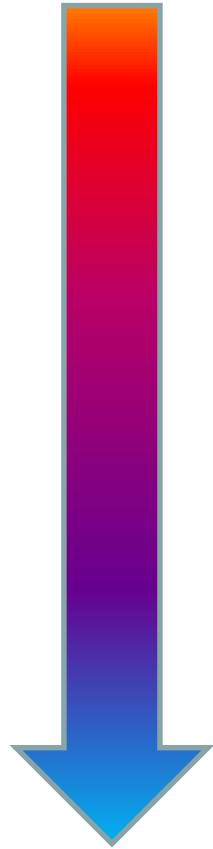
1997 Nobel Prize
Laser cooling and trapping

2001 Nobel Prize
Bose-Einstein
Condensation

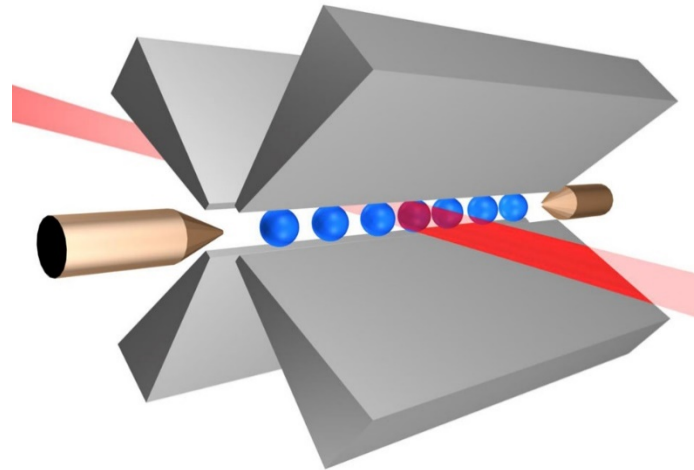
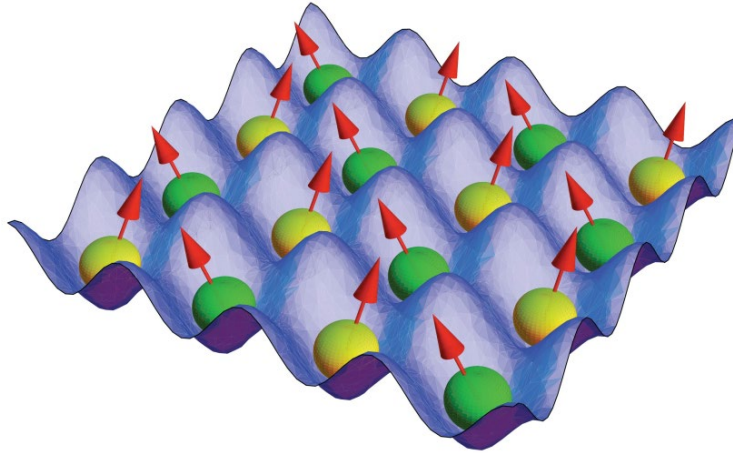
2005 Nobel Prize
Frequency combs

2012 Nobel prize
Quantum control

300K

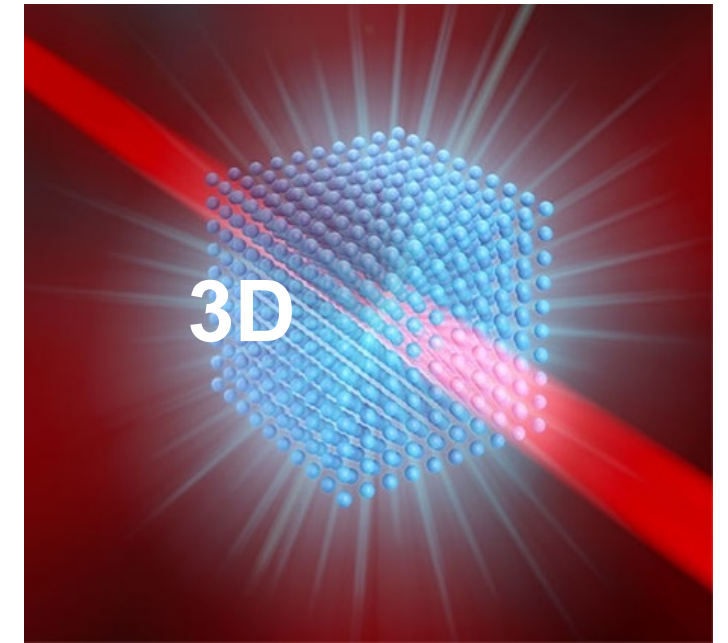


pK



$$\Psi = \left| \begin{array}{cc} -1/2 & +1/2 \\ \uparrow & \downarrow \end{array} \right\rangle + \left| \begin{array}{cc} -5/2 & +5/2 \\ \leftarrow & \rightarrow \end{array} \right\rangle$$

\vec{B}



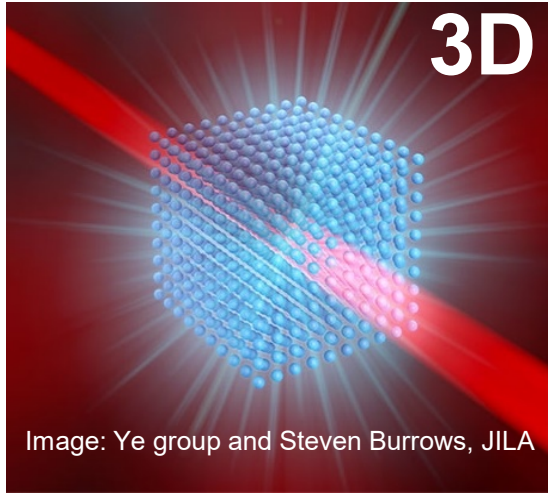
Atoms are now:

Ultracold

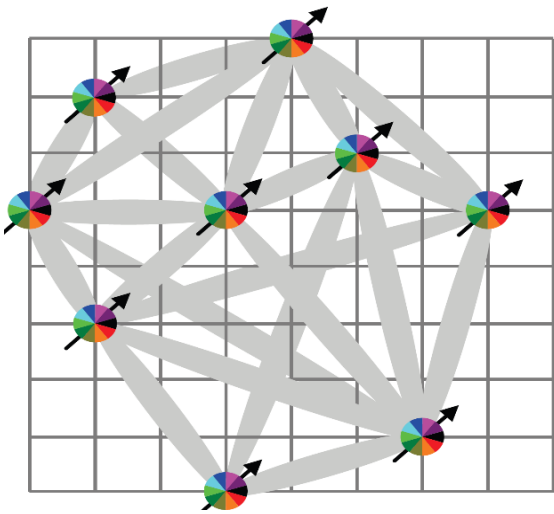
Trapped

Precisely controlled

Numerous applications that need precise atomic data

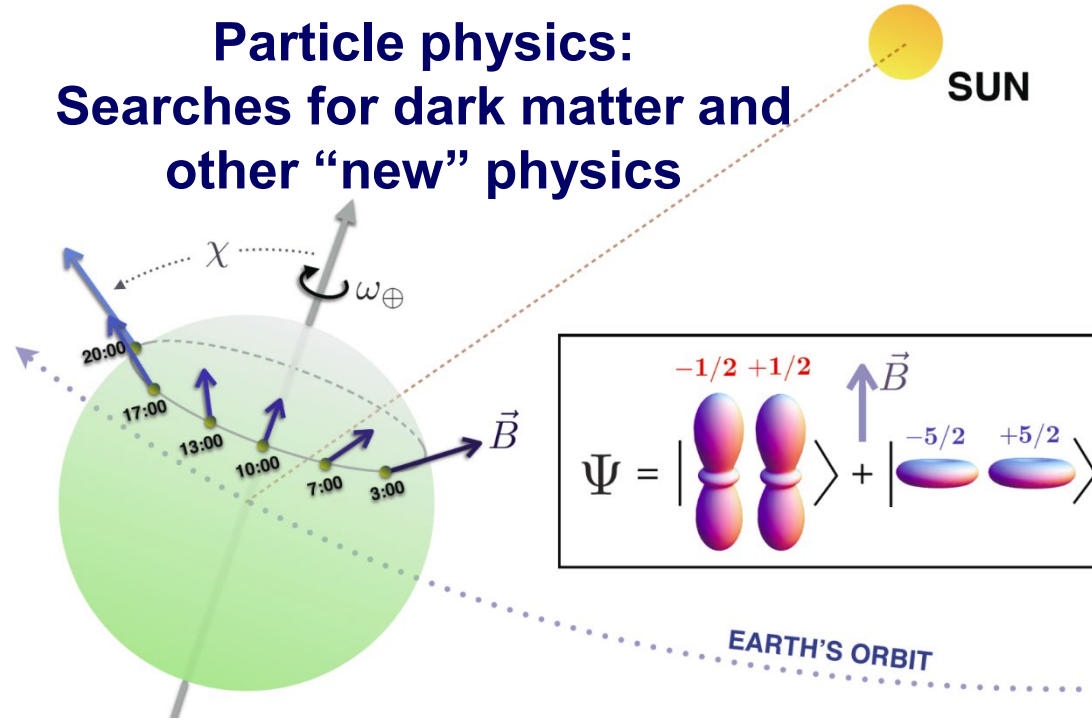


Atomic clocks



Ultracold atoms

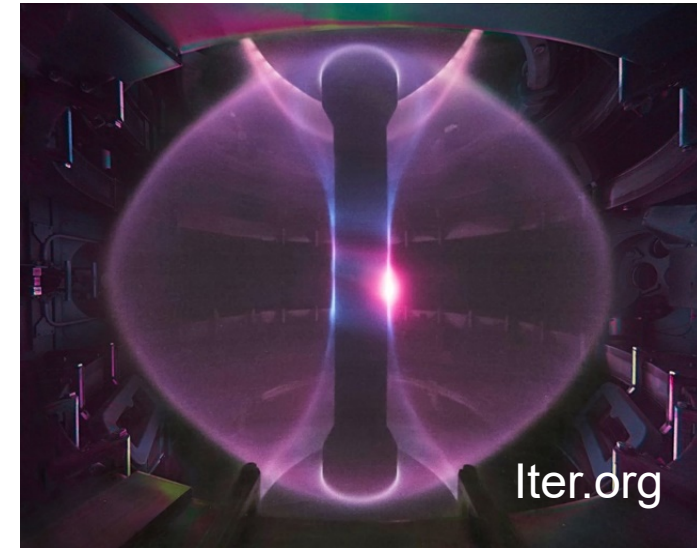
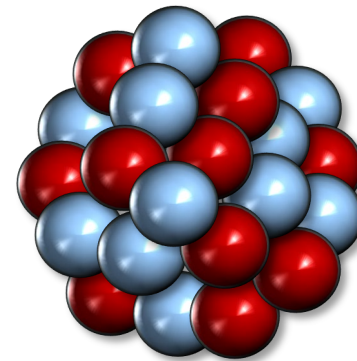
Quantum computing and simulation



Astrophysics



Nuclear and hadronic physics - extracting nuclear properties



Plasma physics

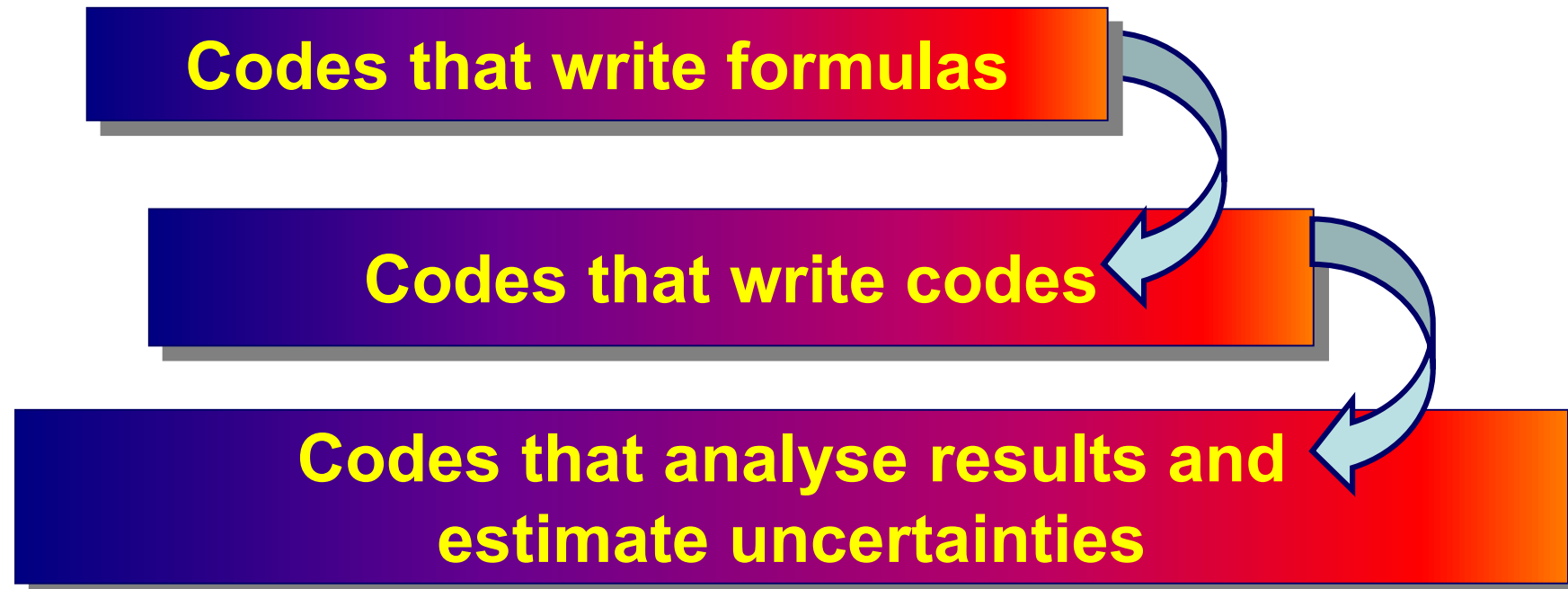
Problems with currently available atomic community codes

- Old - developed initially in 1980s and 1990s, with later updates
- Unsupported or unwieldy (too many updates by many people)
- Designed to produce large volumes of low-precision data
- Poorly documented and/or require expert knowledge to use
- No estimates of how accurate the results are
- Do not serve the need of the present community

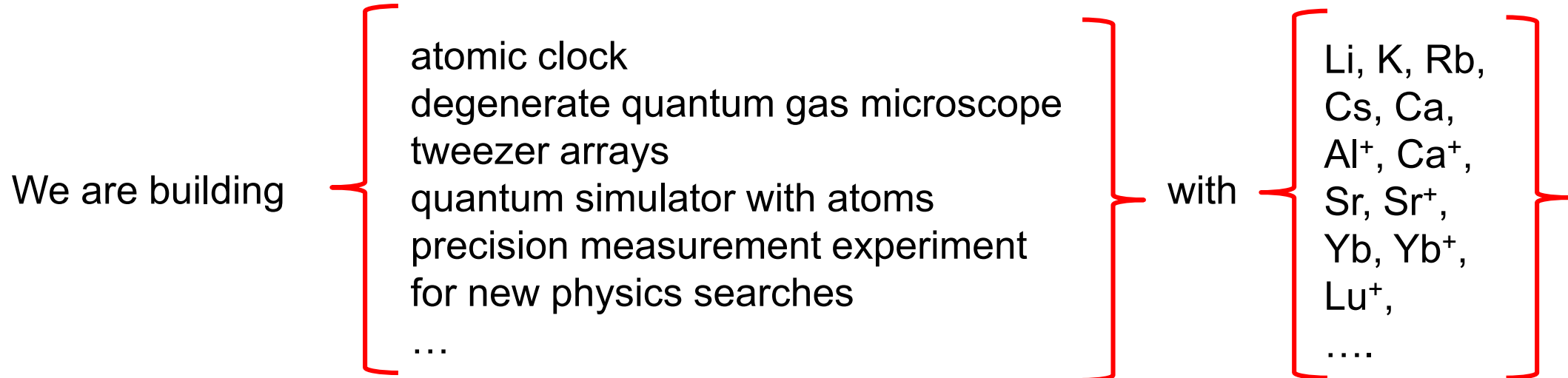
There are very few groups in the world developing new atomic codes

University of Delaware team & collaborators

- We have been developing high precision atomic codes and applying them to solve completely different problems for over 20 years
- All codes are written by us
- Because we have several *ab initio* codes we can estimate how accurate numbers are – we are the only group to routinely publish reliable uncertainties
- Most accurate and versatile set of atomic code packages in the world



Numerous emails from experimental colleagues



We need [*transition rates, branching ratios, lifetimes, polarizabilities, ...*]

We found some data in your papers – will it be possible to provide?

Would you collaborate with us on the interpretation of our measurements?

We have measured ... but the values differ strongly from the existing literature values.

Will it be possible for you to calculate these?

Variations: atoms are missing from the trap, no expected signal observed, ...

**We plan to measure [....]. Will these quantities be useful in testing your new codes?
What else will be useful to measure?**

NSF PIF: Physics at the Information Frontier Program



Contents lists available at [ScienceDirect](#)

Computer Physics Communications

journal homepage: www.elsevier.com/locate/cpc

CI-MBPT: A package of programs for relativistic atomic calculations based on a method combining configuration interaction and many-body perturbation theory[☆]

M.G. Kozlov^{a,b,*}, S.G. Porsev^{a,c,**}, M.S. Safronova^{c,d}, I.I. Tupitsyn^e

Comput. Phys. Commun. 195, 199 (2015).

Used by theory rather than experimental groups

Observations: users, numbers, and codes

- Atomic physics has ~ 90% to 10% ratio of experiment vs. theory
- Very large number of users need **numbers**, preferably with error bars, rather than codes.
- The threshold to download, understand and run a complicated set of codes of high-precision codes without much support is extremely high – usually not done by experimental groups.
- Present high precision codes are complicated and requires expert knowledge to run successfully and access to significant computational resources.
- **To develop even more accurate codes we need precision experimental benchmarks, so we need to support precision experiments!**

PERIODIC TABLE
Atomic Properties of the Elements

NIST

National Institute of Standards and Technology
Technology Administration, U.S. Department of Commerce

There are really a lot of atoms!

- Solids
- Liquids
- Gases
- Artificially Prepared

Group 1 IA												18 VIIIA																	
1	2											13 IIIA	14 IVA	15 VA	16 VIA	17 VIIA	18 VIIIA												
1	2											5	6	7	8	9	10												
3	4											13	14	15	16	17	18												
11	12	3	4	5	6	7	8	9	10	11	12																		
19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36												
37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54												
55	56	Lanthanides										72	73	74	75	76	77	78	79	80	81	82	83	84	85	86			
87	88											Actinides										104	105	106	107	108	109	110	111
57	58	59	60	61	62	63	64	65	66	67	68											69	70	71	89	90	91	92	93
La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu	Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr

Neutral atom: Fe

Ions: keep removing electrons

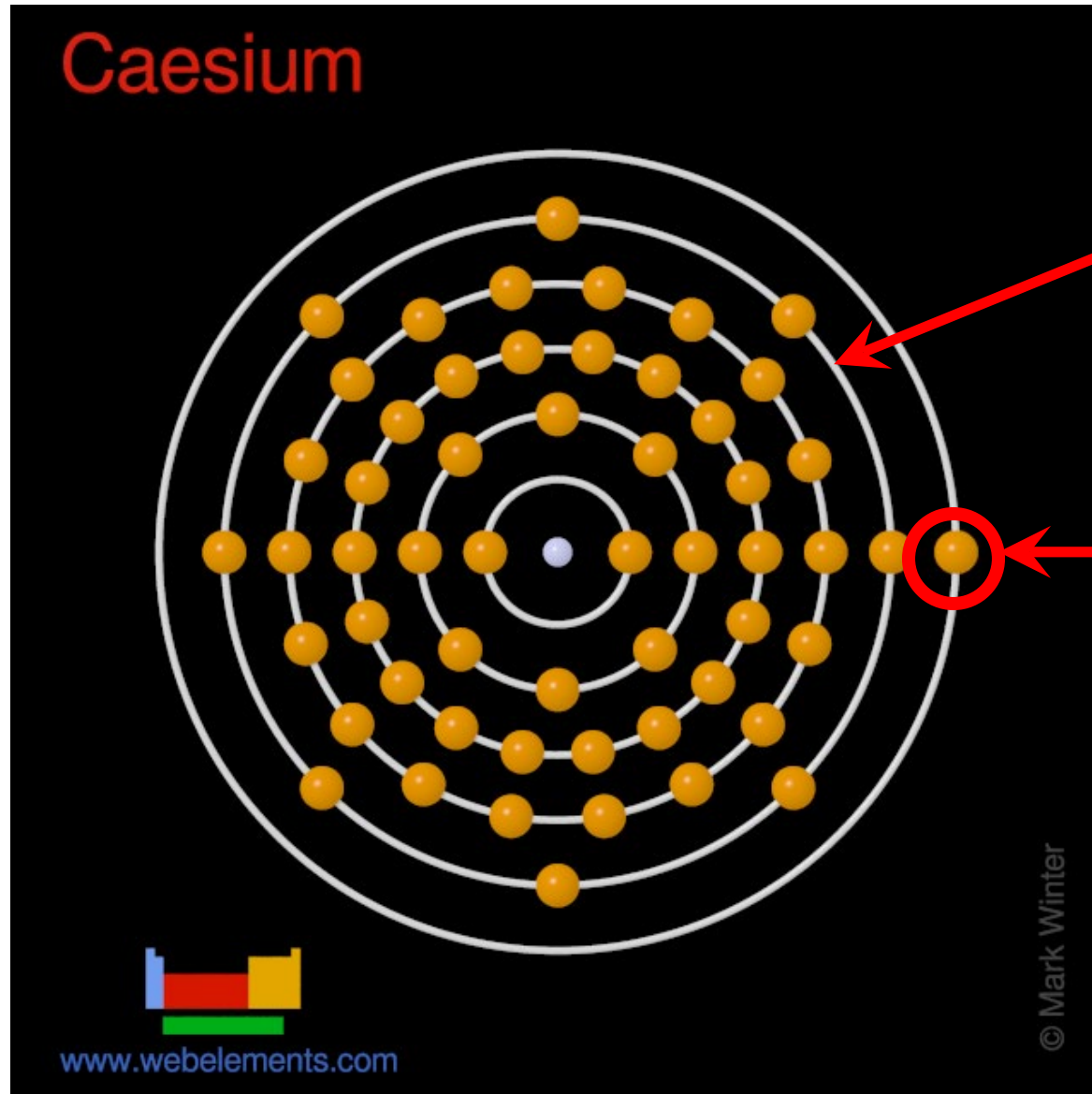
- Fe⁺ Fe¹⁴⁺
- Fe²⁺ Fe¹⁵⁺
- Fe³⁺ Fe¹⁶⁺
- Fe⁴⁺ Fe¹⁷⁺
- Fe⁵⁺ Fe¹⁸⁺
- Fe⁶⁺ Fe¹⁹⁺
- Fe⁷⁺ Fe²⁰⁺
- Fe⁸⁺ Fe²¹⁺
- Fe⁹⁺ Fe²²⁺
- Fe¹⁰⁺ Fe²³⁺
- Fe¹¹⁺ Fe²⁴⁺
- Fe¹²⁺ Fe²⁵⁺
- Fe¹³⁺

[†]Based upon ¹²C. () indicates the mass number of the most stable isotope.

For a description of the data, visit physics.nist.gov/data

NIST SP 966 (September 2003)

Classify atomic calculations by difficulty level



Closed shells

Can be approximated
by a mean field

Single
valence
electron

Classify atomic calculations by difficulty level

3 Li Lithium 6.941 $1s^2 2s^1$ 5.3917	4 Be Beryllium 9.012182 $1s^2 2s^2$ 9.3227
11 Na Sodium 22.989770 [Ne]3s ¹ 5.1391	12 Mg Magnesium 24.3050 [Ne]3s ² 7.6462
19 K Potassium 39.0983 [Ar]4s ¹ 4.3407	20 Ca Calcium 40.078 [Ar]4s ² 6.1132
37 Rb Rubidium 85.4678 [Kr]5s ¹ 4.1771	38 Sr Strontium 87.62 [Kr]5s ² 5.6949
55 Cs Cesium 132.90545 [Xe]6s ¹ 3.8939	56 Ba Barium 137.327 [Xe]6s ² 5.2117
87 Fr Francium (223) [Rn]7s ¹ 4.0727	88 Ra Radium (226) [Rn]7s ² 5.2784

Group 1
Calculations we can do “routinely”, with default parameters

1 – 2(3) valence electrons

Can automate

Group 2
Calculations that require expert knowledge

(3)/4-5 valence electrons or special cases with more valence electrons

Only calculations of wave functions requires expert knowledge

Group 3
No precision methods exist: **exponential scaling** with the number of valence electrons

60 Nd Neodymium 144.24 [Xe]4f ⁴ 6s ² 5.5250	61 Pm Promethium (145) [Xe]4f ⁵ 6s ² 5.582	62 Sm Samarium 150.36 [Xe]4f ⁶ 6s ² 5.6437	63 Eu Europium 151.964 [Xe]4f ⁷ 6s ² 5.6704	64 Gd Gadolinium 157.25 [Xe]4f ⁷ 5d ¹ 6s ² 6.1498	65 Tb Terbium 158.92534 [Xe]4f ⁹ 6s ² 5.8538	66 Dy Dysprosium 162.500 [Xe]4f ¹⁰ 6s ² 5.9389	67 Ho Holmium 164.93032 [Xe]4f ¹¹ 6s ² 6.0215	68 Er Erbium 167.259 [Xe]4f ¹² 6s ² 6.1077	69 Tm Thulium 168.93421 [Xe]4f ¹³ 6s ² 6.1843
92 U Uranium 238.02891 [Rn]5f ³ 6d ¹ 7s ² 6.1941	93 Np Neptunium (237) [Rn]5f ⁴ 6d ¹ 7s ² 6.2657	94 Pu Plutonium (244) [Rn]5f ⁶ 7s ² 6.0260	95 Am Americium (243) [Rn]5f ⁷ 7s ² 5.9738	96 Cm Curium (247) [Rn]5f ⁷ 6d ¹ 7s ² 5.9914	97 Bk Berkelium (247) [Rn]5f ⁹ 7s ² 6.1979	98 Cf Californium (251) [Rn]5f ¹⁰ 7s ² 6.2817	99 Es Einsteinium (252) [Rn]5f ¹¹ 7s ² 6.42	100 Fm Fermium (257) [Rn]5f ¹² 7s ² 6.50	101 Md Mendelevium (258) [Rn]5f ¹³ 7s ² 6.58

Half-filled shells and holes in shells

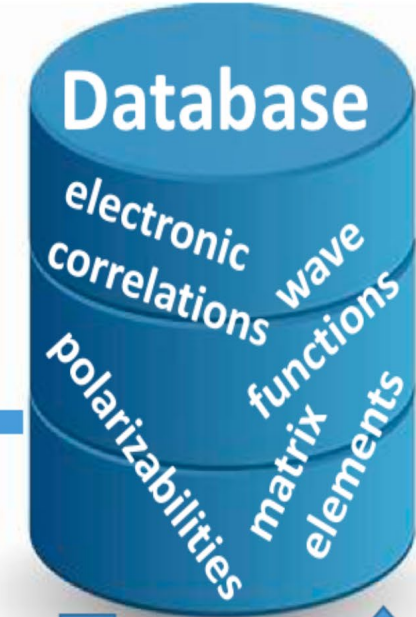
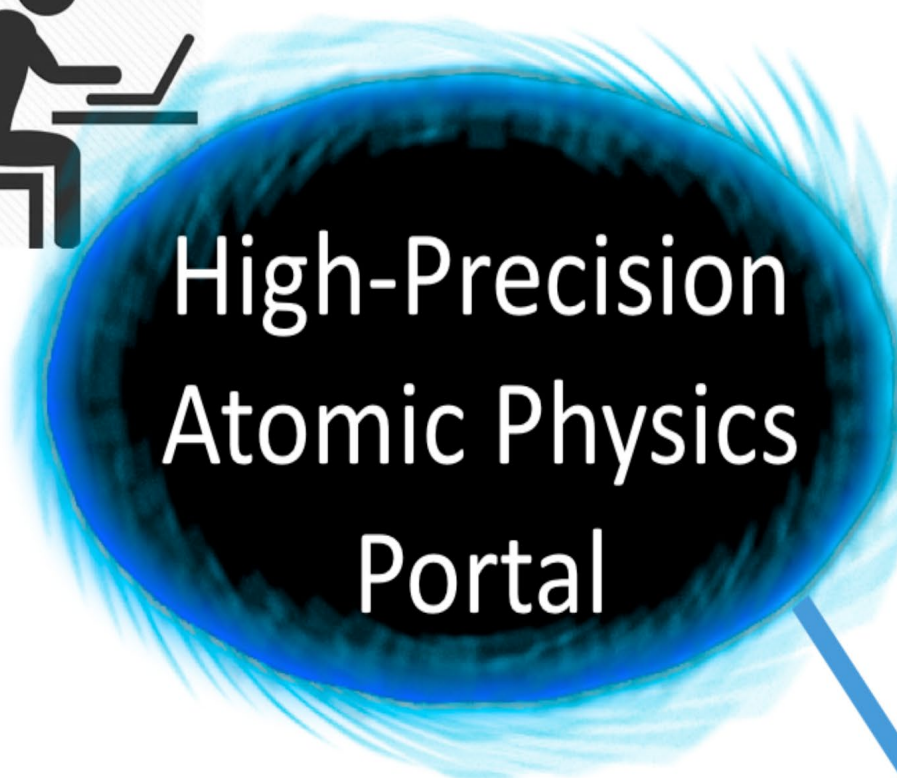
Method development in progress, need new ideas – machine learning

25 Mn Manganese 54.938049 [Ar]3d ⁵ 4s ² 7.4340	26 Fe Iron 55.845 [Ar]3d ⁶ 4s ² 7.9024	27 Co Cobalt 58.933200 [Ar]3d ⁷ 4s ² 7.8810	28 Ni Nickel 58.6934 [Ar]3d ⁸ 4s ² 7.6388
43 Tc Technetium (98) [Kr]4d ⁵ 5s ² 7.28	44 Ru Ruthenium 101.07 [Kr]4d ⁷ 5s ¹ 7.3605	45 Rh Rhodium 102.90550 [Kr]4d ⁸ 5s ¹ 7.4589	46 Pd Palladium 106.42 [Kr]4d ¹⁰ 8.3369
75 Re Rhenium 186.207 [Xe]4f ¹⁴ 5d ⁵ 6s ² 7.8335	76 Os Osmium 190.23 [Xe]4f ¹⁴ 5d ⁶ 6s ² 8.4382	77 Ir Iridium 192.217 [Xe]4f ¹⁴ 5d ⁷ 6s ² 8.9670	78 Pt Platinum 195.078 [Xe]4f ¹⁴ 5d ⁹ 6s ¹ 8.9588

Community – driven project: there is enormous need for data

Applications in science and engineering

- quantum information
- degenerate quantum gases
- atomic clocks
- precision measurements
- plasma physics
- astrophysics
- studies of fundamental physics



on-demand execution

automatic ingest

Atomic Physics computational codes

Difficulty Groups 1 and 2

Building on:

- CI+MBPT/CI+all-order program package and expertise
- Portal technology (Science Gateways, Hubzero,...)
- Parallel programming methodology

COMPUTER, CALCULATE!

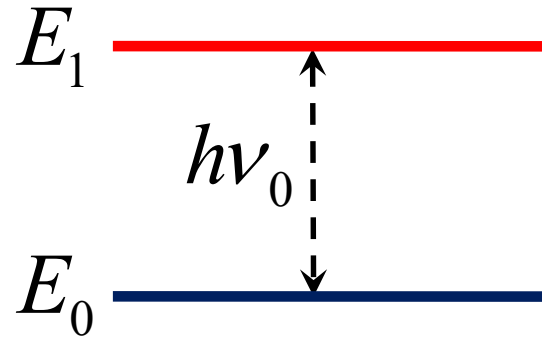
To boldly go where no one has gone before ...



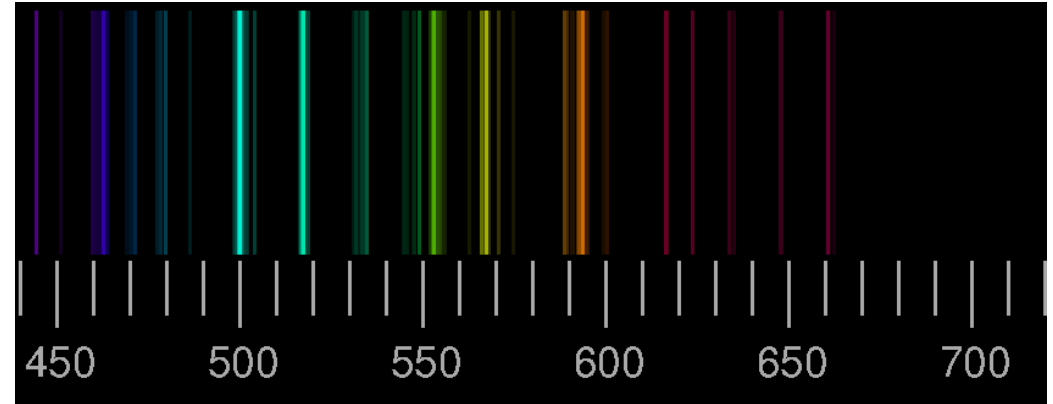
www.film.ru

How to serve the most diverse group of users?

Most requested data:
transition matrix
elements and
polarizabilities

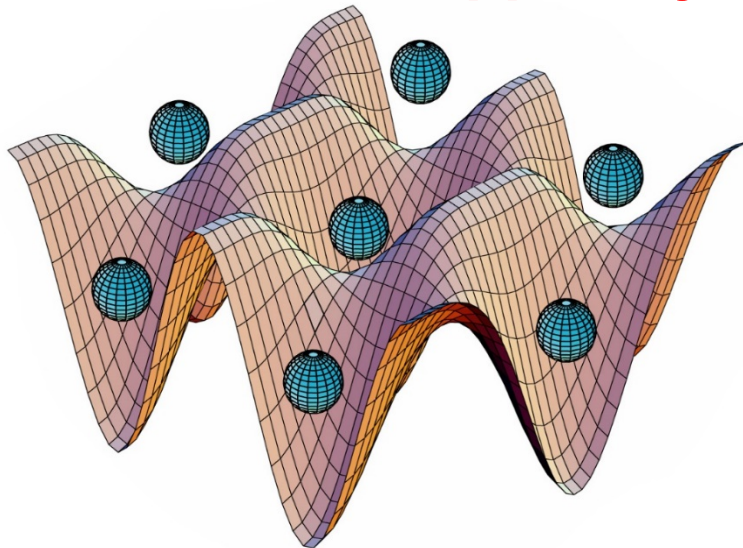


What is transition probability?



What is transition energy?

Atoms are now **trapped by light**



Need electric-dipole polarizability
 α to determine how deep the trap
will be for specified laser
wavelength

$$U \propto \alpha(\lambda)$$

How to serve the most diverse group of users?

Most requested data: transition matrix elements and polarizabilities will be pre-calculated for atoms/ions of most interest, Group 1 and some Group 2.

Uncertainty estimates will be provided for all data.

This will require vast computations so the code packages are being completely automated for such data production for Group 1 atoms/ions.

Users who need other data for these systems: all wave functions from runs above will be stored so other data can be requested – will be calculated automatically. Users do not need to know anything about codes.

Advanced users – frequent need of data and theory groups

All codes will be released to public – optimized and **very** user friendly.

We will have tutorials and workshops providing training to use the codes.

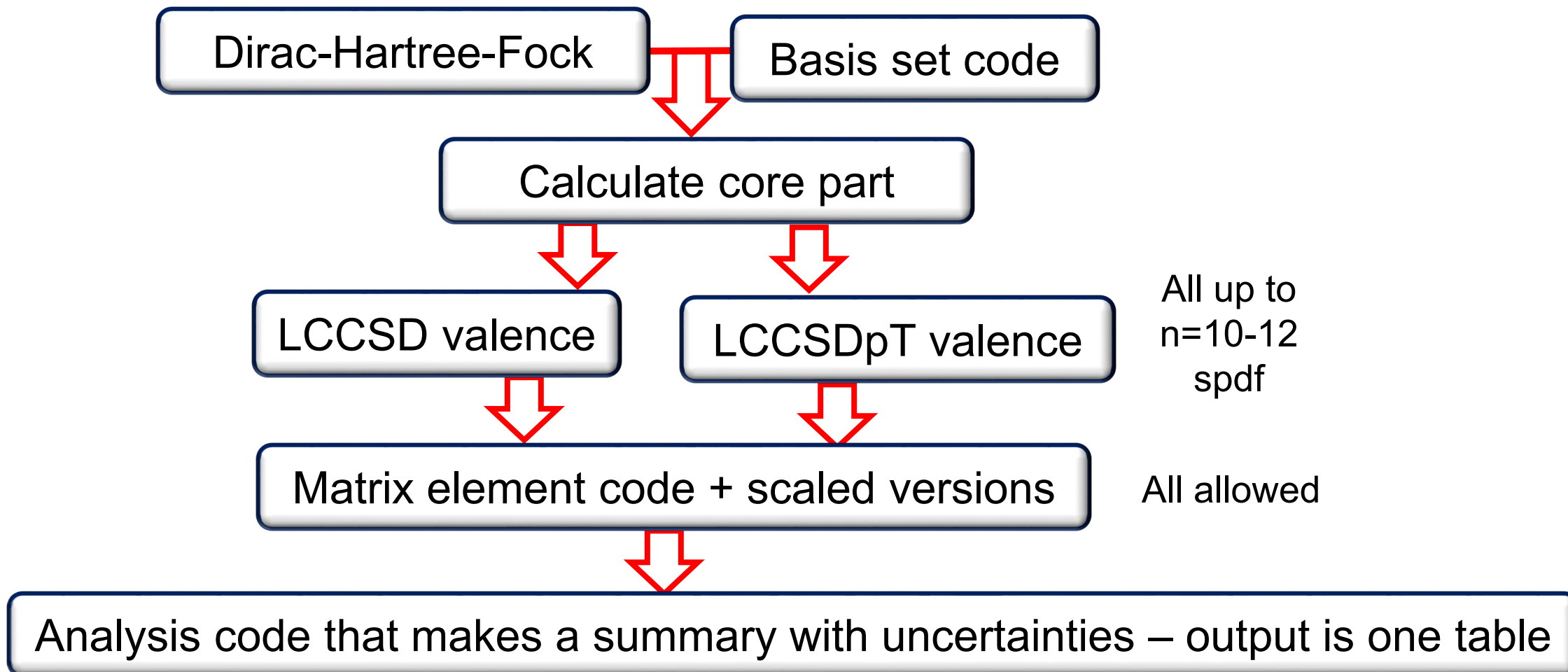
Other groups will send us representatives for several months to train as experts

Population of the database will be completely automated.

Example: monovalent systems

Present data are incomplete and scattered through the projects

COMPUTER, CALCULATE CS!



Final code output: transition E1 matrix elements in atomic units

Transition	DHF	SD	SDsc	SDpT	SDpTsc	Final
$3d_{3/2} - 4f_{5/2}$	2.6059	1.8660	1.9265	1.9265	1.9051	1.93(2)
$3d_{3/2} - 4p_{1/2}$	3.0825	2.4173	2.4636	2.4677	2.4503	2.46(1)
$3d_{3/2} - 4p_{3/2}$	1.3764	1.0788	1.0996	1.1014	1.0937	1.100(6)
$3d_{3/2} - 5f_{5/2}$	1.5216	1.1655	1.1917	1.1937	1.1846	1.192(7)
$3d_{3/2} - 5p_{1/2}$	0.0063	0.1254	0.1055	0.1097	0.1154	0.11(2)
$3d_{3/2} - 5p_{3/2}$	0.0008	0.0590	0.0501	0.0521	0.0545	0.050(9)
$3d_{3/2} - 6f_{5/2}$	1.0430	0.8195	0.8347	0.8366	0.8313	0.83(2)
$3d_{3/2} - 6p_{1/2}$	0.0276	0.0683	0.0607	0.0623	0.0640	0.061(8)
$3d_{3/2} - 6p_{3/2}$	0.0138	0.0319	0.0285	0.0293	0.0300	0.028(3)
$3d_{3/2} - 7f_{5/2}$	0.7794	0.6201	0.6303	0.6319	0.6283	0.63(1)
$3d_{3/2} - 7p_{1/2}$	0.0266	0.0464	0.0418	0.0429	0.0436	0.042(7)
$3d_{3/2} - 7p_{3/2}$	0.0128	0.0217	0.0196	0.0201	0.0204	0.020(3)
$3d_{3/2} - 8f_{5/2}$	0.6142	0.4921	0.4996	0.5009	0.4982	0.500(8)
$3d_{3/2} - 8p_{1/2}$	0.0227	0.0346		0.0322		0.032(3)
$3d_{3/2} - 8p_{3/2}$	0.0108	0.0161		0.0151		0.015(2)
$3d_{3/2} - 9f_{5/2}$	0.5017	0.4037	0.4096	0.4107	0.4086	0.410(6)
$3d_{3/2} - 9p_{1/2}$	0.0191	0.0272		0.0254		0.025(3)
$3d_{3/2} - 9p_{3/2}$	0.0090	0.0127		0.0119		0.012(1)
$3d_{5/2} - 10f_{5/2}$	0.1126	0.0910	0.0922	0.0925	0.0920	0.092(1)
$3d_{5/2} - 10f_{7/2}$	0.5035	0.4068	0.4124	0.4136	0.4115	0.412(6)

Li	Na	K	Rb	Cs	Fr
Be ⁺	Mg ⁺	Ca ⁺	Sr ⁺	Ba ⁺	Ra ⁺
Be	Mg	Ca	Sr	Ba	Ra
B ⁺	Al ⁺	In ⁺	Lu ⁺	Tl ⁺	Th ³⁺
Cd	In	Lu	Tl	Pb	Th

Li	Na	K	Rb	Cs	Fr
Be ⁺	Mg ⁺	Ca ⁺	Sr ⁺	Ba ⁺	Ra ⁺
Be	Mg	Ca	Sr	Ba	Ra
B ⁺	Al ⁺	In ⁺	Lu ⁺	Tl ⁺	Th ³⁺
Cd	In	Lu	Tl	Pb	Th

Cs

All data

6s	6p _{1/2}	6p _{3/2}	5d _{3/2}	5d _{5/2}	4f _{5/2}	4f _{7/2}
7s	7p _{1/2}	7p _{3/2}	6d _{3/2}	6d _{5/2}	5f _{5/2}	5f _{7/2}
8s	8p _{1/2}	8p _{3/2}	7d _{3/2}	7d _{5/2}	6f _{5/2}	6f _{7/2}
9s	9p _{1/2}	9p _{3/2}	8d _{3/2}	8d _{5/2}	7f _{5/2}	7f _{7/2}
10s	10p _{1/2}	10p _{3/2}	9d _{3/2}	9d _{5/2}	8f _{5/2}	8f _{7/2}
11s	11p _{1/2}	11p _{3/2}	10d _{3/2}	10d _{5/2}	9f _{5/2}	9f _{7/2}
12s	12p _{1/2}	12p _{3/2}	11d _{3/2}	11d _{5/2}	10f _{5/2}	10f _{7/2}

Cs

All data

6s	6p _{1/2}	6p _{3/2}	5d _{3/2}	5d _{5/2}	4f _{5/2}	4f _{7/2}
7s	7p _{1/2}	7p _{3/2}	6d _{3/2}	6d _{5/2}	5f _{5/2}	5f _{7/2}
8s	8p _{1/2}	8p _{3/2}	7d _{3/2}	7d _{5/2}	6f _{5/2}	6f _{7/2}
9s	9p _{1/2}	9p _{3/2}	8d _{3/2}	8d _{5/2}	7f _{5/2}	7f _{7/2}
10s	10p _{1/2}	10p _{3/2}	9d _{3/2}	9d _{5/2}	8f _{5/2}	8f _{7/2}
11s	11p _{1/2}	11p _{3/2}	10d _{3/2}	10d _{5/2}	9f _{5/2}	9f _{7/2}
12s	12p _{1/2}	12p _{3/2}	11d _{3/2}	11d _{5/2}	10f _{5/2}	10f _{7/2}

Output: table of electric-dipole matrix elements

Print or download in Excel format

Transition rates, branching ratios and lifetime options will be added as well.

6p _{3/2}	6s _{1/2}	6.38(8)	6p _{3/2}	5d _{3/2}	3.19(7)	6p _{3/2}	5d _{5/2}	9.7(2)
6p _{3/2}	7s _{1/2}	6.48(2)	6p _{3/2}	6d _{3/2}	2.09(3)	6p _{3/2}	6d _{5/2}	6.13(9)
6p _{3/2}	8s _{1/2}	1.46(2)	6p _{3/2}	7d _{3/2}	0.976(0)	6p _{3/2}	7d _{5/2}	2.89(3)
6p _{3/2}	9s _{1/2}	0.766(9)	6p _{3/2}	8d _{3/2}	0.607(8)	6p _{3/2}	8d _{5/2}	1.81(2)
6p _{3/2}	10s _{1/2}	0.505(6)	6p _{3/2}	9d _{3/2}	0.430(6)	6p _{3/2}	9d _{5/2}	1.28(2)
6p _{3/2}	11s _{1/2}	0.370(4)	6p _{3/2}	10d _{3/2}	0.328(5)	6p _{3/2}	10d _{5/2}	0.979(6)
6p _{3/2}	12s _{1/2}	0.289(3)	6p _{3/2}	11d _{3/2}	0.262(4)	6p _{3/2}	11d _{5/2}	0.782(5)
6p _{3/2}	13s _{1/2}	0.235(3)	6p _{3/2}	12d _{3/2}	0.2201	6p _{3/2}	12d _{5/2}	0.6585

Uncertainties are given in parenthesis.

High-precision experimental data will be provided where available with references.
The goal of the portal is to provide recommended data.

Cs

Other properties not in database

6s	6p _{1/2}	6p _{3/2}	5d _{3/2}	5d _{5/2}	4f _{5/2}	4f _{7/2}
7s	7p _{1/2}	7p _{3/2}	6d _{3/2}	6d _{5/2}	5f _{5/2}	5f _{7/2}
8s	8p _{1/2}	8p _{3/2}	7d _{3/2}	7d _{5/2}	6f _{5/2}	6f _{7/2}
9s	9p _{1/2}	9p _{3/2}	8d _{3/2}	8d _{5/2}	7f _{5/2}	7f _{7/2}
10s	10p _{1/2}	10p _{3/2}	9d _{3/2}	9d _{5/2}	8f _{5/2}	8f _{7/2}
11s	11p _{1/2}	11p _{3/2}	10d _{3/2}	10d _{5/2}	9f _{5/2}	9f _{7/2}
12s	12p _{1/2}	12p _{3/2}	11d _{3/2}	11d _{5/2}	10f _{5/2}	10f _{7/2}

- E2, E3, M1, M2, M3 transition matrix elements
- A and B hyperfine constants
- Parity-violating matrix element
- T-odd matrix element
- Lorentz violating matrix elements

- Click on 1 or 2 states (depends on a property)
- Select needed property from the pull-down menu – it will be computed automatically using pre-stored wave functions

Polarizability portal page

User will click on element

Sr

Then select a state from the list

$5s^2 \ ^1S_0$

Could also select another state to get a magic wavelength (where two curves cross)

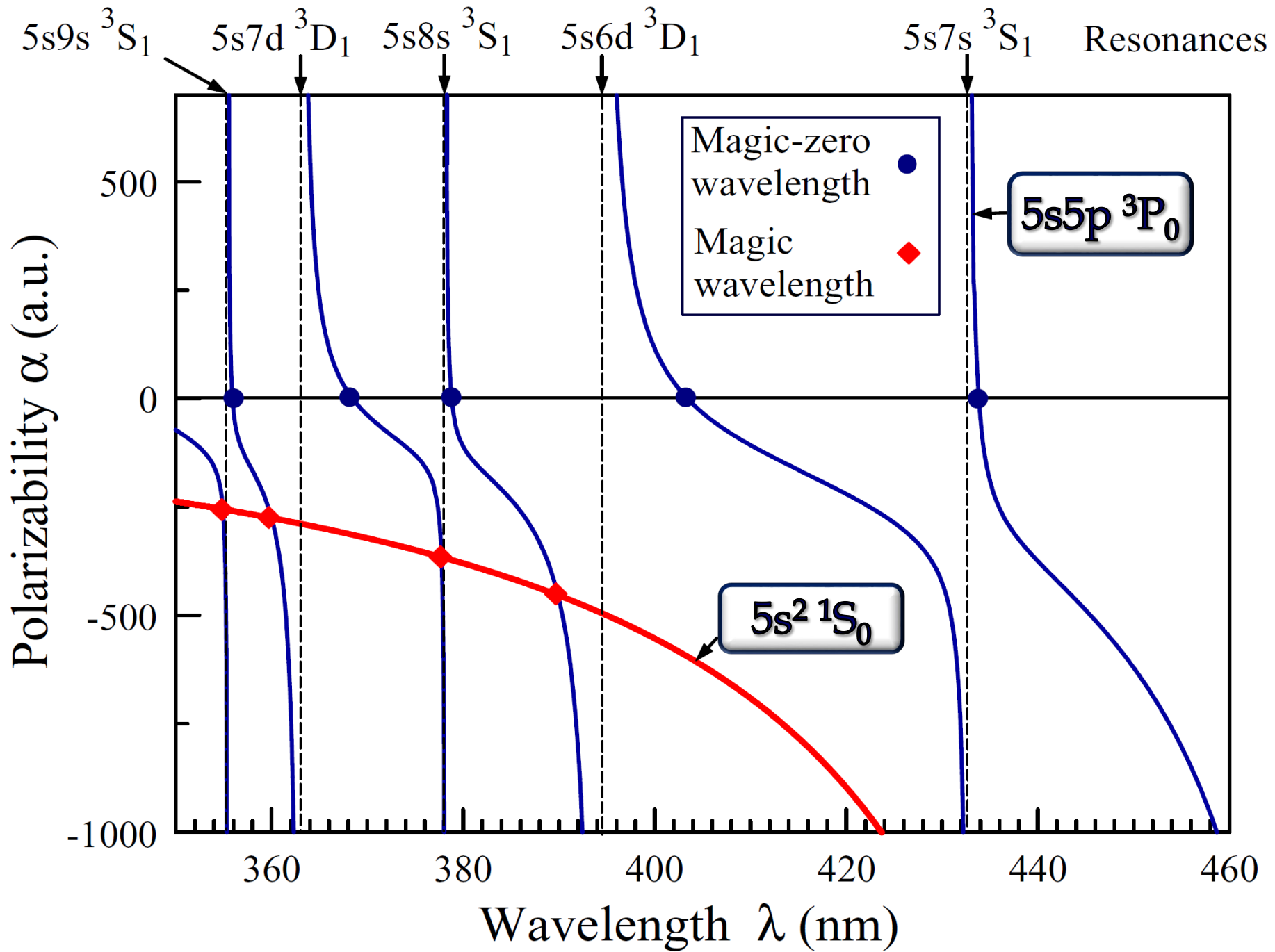
$5s5p \ ^3P_0$

User will enter wavelength range or select a static option

Static

350 - 460 nm

Sr



Summary

Online portal 3-year project started in October 2019



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It is extremely useful for physicists to collaborate with computer scientists!

First version will be online for trial users by DAMOP meeting (June 1, 2020)

Continuing method and code development

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Ilya Tupitsyn (University of St. Petersburg, Russia)

International collaboration will be established to maintain the portal beyond the 3-year project

Our vision: data for the entire periodic table accessible through the portal